## Ionic Compounds

Combine each of the cations (+ion) in the rows with each anion (-ion) in the columns to create the correct chemical formula.

	I <sup>-1</sup>	NO <sub>3</sub> -1	S <sup>-2</sup>	PO <sub>4</sub> -3	O <sup>-2</sup>
Na <sup>+1</sup>	NaI				
Cu <sup>+2</sup>	CuI <sub>2</sub>				
Fe <sup>+3</sup>					
Fe <sup>+2</sup>					
$\mathrm{NH_4}^{+1}$					
Sn <sup>+4</sup>					
$Al^{+3}$					