## Mass - Mass Problems

Using the given mass, figure out the masses of the remaining products and reactants in the following chemical equations. Be sure the equations are balanced before you attempt the problem. Show your work to receive credit!

1. $\mathrm{H}_{2} \mathrm{O}_{2} \longrightarrow \mathrm{H}_{2} \mathrm{O}+\mathrm{O}_{2}$

Given $128 \mathrm{~g} \mathrm{O}_{2}$
2. $\mathrm{Pb}\left(\mathrm{NO}_{3}\right)_{2}+\mathrm{NaI} \longrightarrow \quad \mathrm{PbI}_{2}+\quad \mathrm{NaNO}_{3}$

Given 50.0 g NaI
3. $\mathrm{AgNO}_{3}+\mathrm{Cu} \longrightarrow \mathrm{Cu}\left(\mathrm{NO}_{3}\right)_{2}+\mathrm{Ag}$ Given 127 g Cu
4. $\mathrm{Cu}+\mathrm{S} \longrightarrow \mathrm{Cu}_{2} \mathrm{~S}$

Given 32.0 g S
5. $\mathrm{Zn}+\mathrm{HCl} \longrightarrow \quad \mathrm{ZnCl}_{2}+\mathrm{H}_{2}$

Given $10.0 \mathrm{~g} \mathrm{H}_{2}$

