## Charles' Law Problems

Name:\_\_\_\_\_

Correct the following volumes of gases for the temperature changes indicated. Assume the pressure is constant.

1. 2.90 m<sup>3</sup> at 226K to 23°C

2.  $376 \text{ cm}^3 \text{ at } 379 \text{K to } 51^{\circ} \text{C}$ 

3. 7.91 L at 52°C to 538K

4. 667 mL at 431K to 41°C

5. 4.82 ft<sup>3</sup> at 22°C to 31°C