## Charles' Law Problems

Name: $\qquad$

Correct the following volumes of gases for the temperature changes indicated. Assume the pressure is constant.

1. $2.90 \mathrm{~m}^{3}$ at 226 K to $23^{\circ} \mathrm{C}$
2. $376 \mathrm{~cm}^{3}$ at 379 K to $51^{\circ} \mathrm{C}$
3. 7.91 L at $52^{\circ} \mathrm{C}$ to 538 K
4. 667 mL at 431 K to $41^{\circ} \mathrm{C}$
5. $4.82 \mathrm{ft}^{3}$ at $22^{\circ} \mathrm{C}$ to $31^{\circ} \mathrm{C}$
