

I can...

- _____ 1. Define the terms “substance” and “mixture”.
- _____ 2. Distinguish between substances and mixtures based on observations and descriptions.
- _____ 3. Define the terms “element” and “compound”.
- _____ 4. Distinguish between elements and compounds based on chemical formulas. (*or by using the periodic table.*)
- _____ 5. Compare and contrast heterogeneous mixtures and homogeneous mixtures.
- _____ 6. Compare and contrast solutions, colloids, and suspensions.
- _____ 7. Classify mixtures as solutions, colloids, or suspensions using information in Table 1, page 456. (*This may be difficult based upon only the scattering of light to separate solutions and colloids, how were you planning on doing this?*)
- _____ 8. Compare and contrast physical and chemical properties. (*Identify physical and chemical properties of matter.*)
- _____ 9. Identify and separate substances using physical properties.
- _____ 10. Compare and contrast physical and chemical changes.
- _____ 11. Identify chemical changes.
- _____ 12. Explain how the law of conservation of mass applies to chemical changes. (*and physical changes.*)